



European
Commission

Horizon 2020
European Union funding
for Research & Innovation



PIONEER
PLASMA CATALYSIS CO₂ RECYCLING



AGH UNIVERSITY OF SCIENCE
AND TECHNOLOGY



UNIVERSITÀ
DI TRENTO

Efficient catalysts for the plasma-assisted Dry Reforming of Methane

Marzia Faedda

Physics PhD Workshop, 2 Dec 2020



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Outline

- **My background**
- **Novel catalysts for plasma-catalytic DRM**
- **Why plasma + catalysis?**
- **The catalysts**
- **The reactor setup**
- **An eclectic work**

My background

2012

BSc in Physics

University of Cagliari

2016

MSc in Physics of Advanced Technologies

University of Torino

Thesis: "N-doped carbon-based catalysts for O₂ and CO₂ electrochemical reduction" @IIT & PoliTo

2019

1st October

PhD – PIONEER

Plasma catalysis for CO₂ recycling

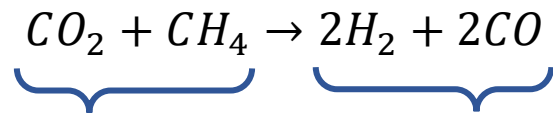
Supervisor	Prof. Monika Motak	AGH - UST in Krakow
Co-supervisor	Prof. Paolo Tosi	University of Trento

Project: Novel catalysts for the plasma-assisted
Dry Reforming of Methane(DRM)



Novel catalysts for plasma-catalytic DRM

Dry Reforming of methane



Two of the most abundant and dangerous greenhouse gases

Synthesis gas
(H₂/CO=1)

$$\underbrace{\Delta H^\circ_{298K} = +247 \text{ kJ mol}^{-1}}_{\text{Highly endothermic reaction}}$$

Highly endothermic reaction

→ High activation temperatures
(>600° C) are necessary

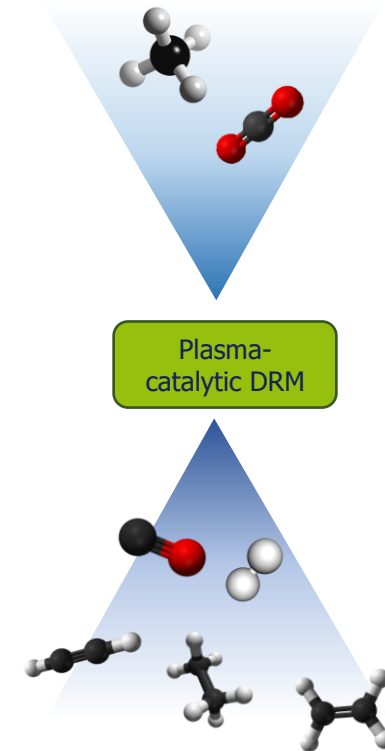
Electrical
discharge
(NRP plasma)

+

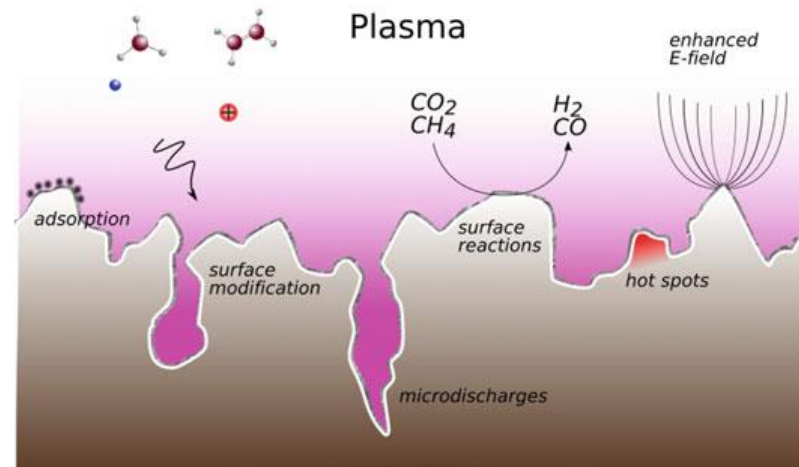
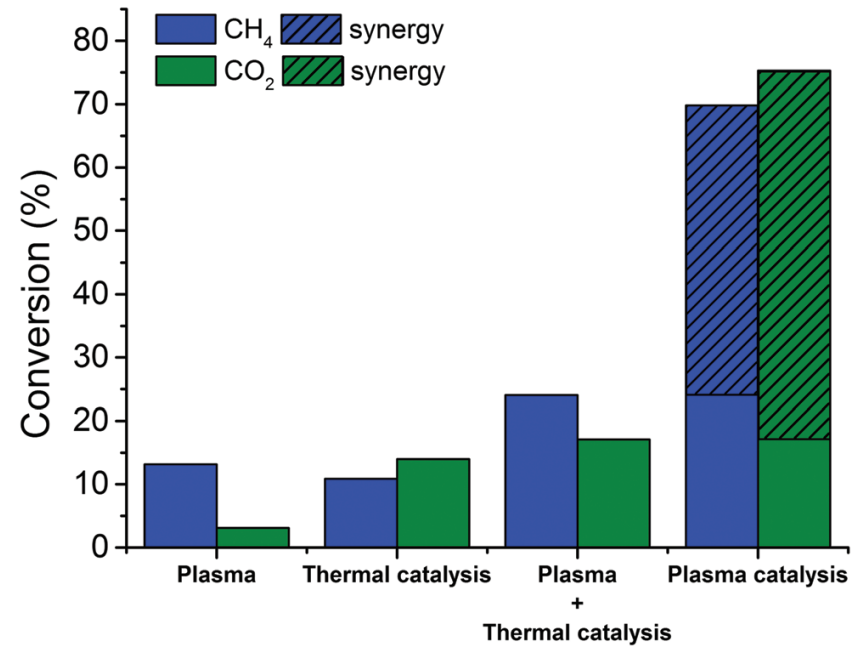
Catalyst



Lower activation energy
Increase energy efficiency
Increase conversion
Control products selectivity



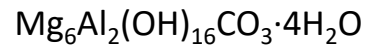
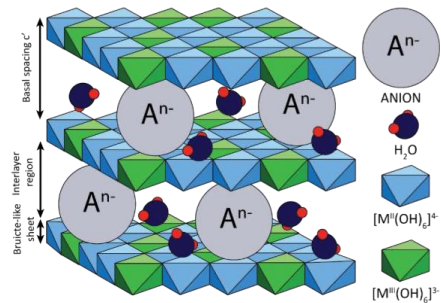
Why plasma + catalysis?



The catalysts

Ni-based hydrotalcite-derived catalyst

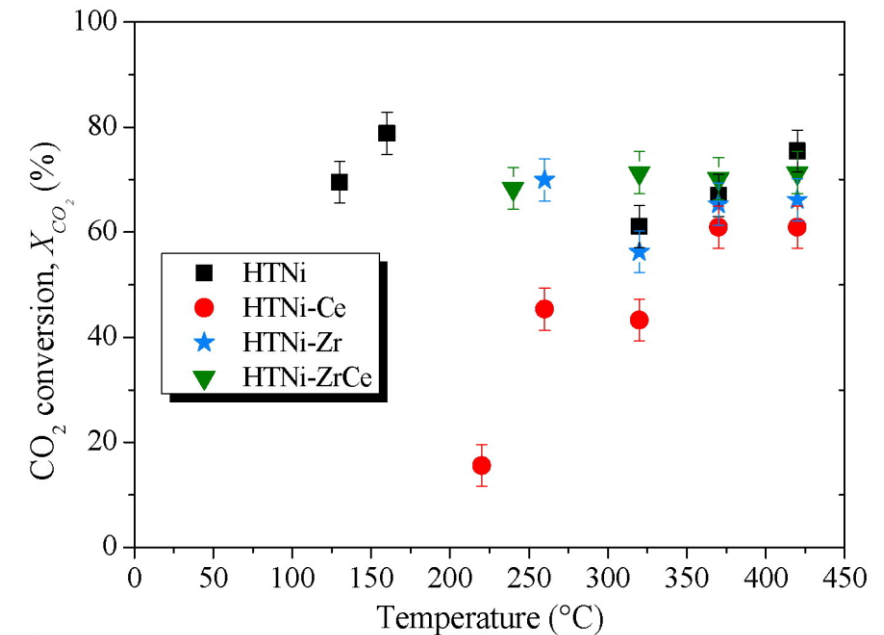
hydrotalcite-like structure



Active material: Nickel

- their surface basicity allows great CO_2 adsorption capability
- hydrotalcite-derived catalysts have been reported to be highly active and stable in DRM.
- high surface area to volume ratio, high porosity
- low cost
- environment-friendly

Debek 2015 –Low T methanation in a DBD plasma



The catalysts

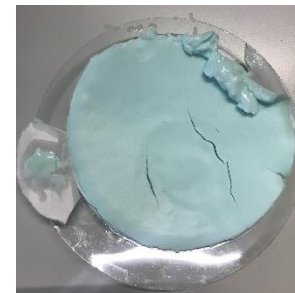
Samples nomenclature

Active metal: Ni	Support : Hydrotalcite
5 wt.%	5NiHT(calc, red, DRM)
10 wt.%	10NiHT(calc, red, DRM)
40 wt.%	40NiHT(calc, red, DRM)

1) Synthesis: co-precipitation method at constant pH and temperature



2) Washing with warm distilled water and filtering



3) Drying overnight at 80° C and grinding into a fine powder



4) Calcination at 550° C for 4 h



Calcined powder



5) Reduction at 900° C for 1 h
5% H₂/Ar flow
150 sccm
UniTn Materials Eng. Dep.



X-Ray Diffractometry

Sample: 40NiHTred

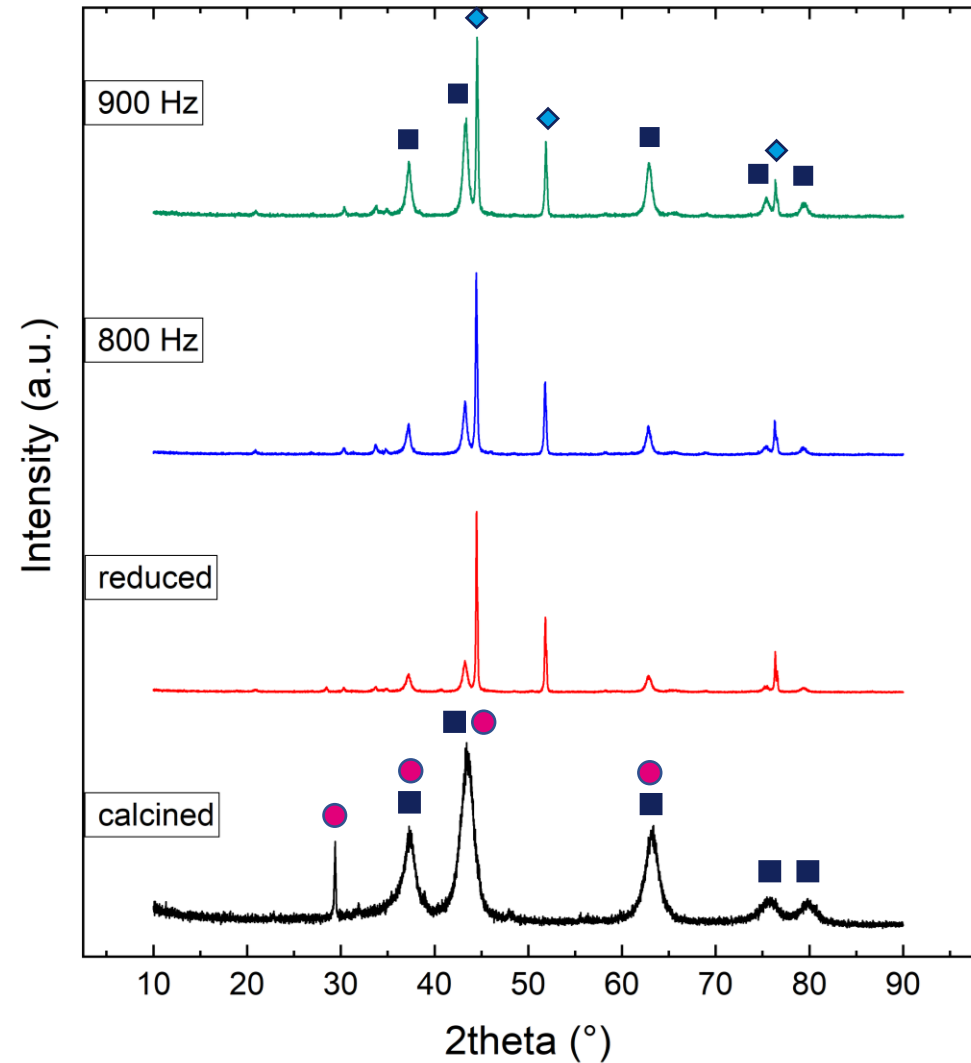
Catalyst crystalline structure after:

- **Preparation steps** (calcination, reduction);
- **DRM reaction**

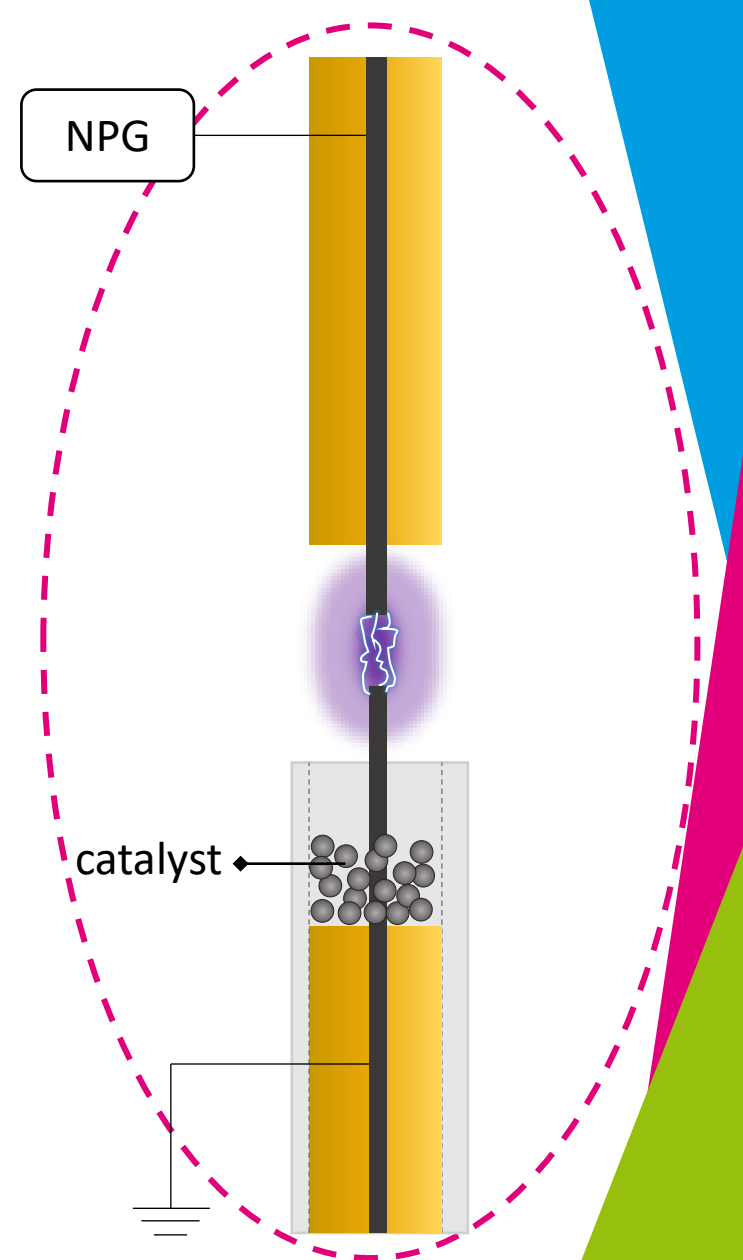
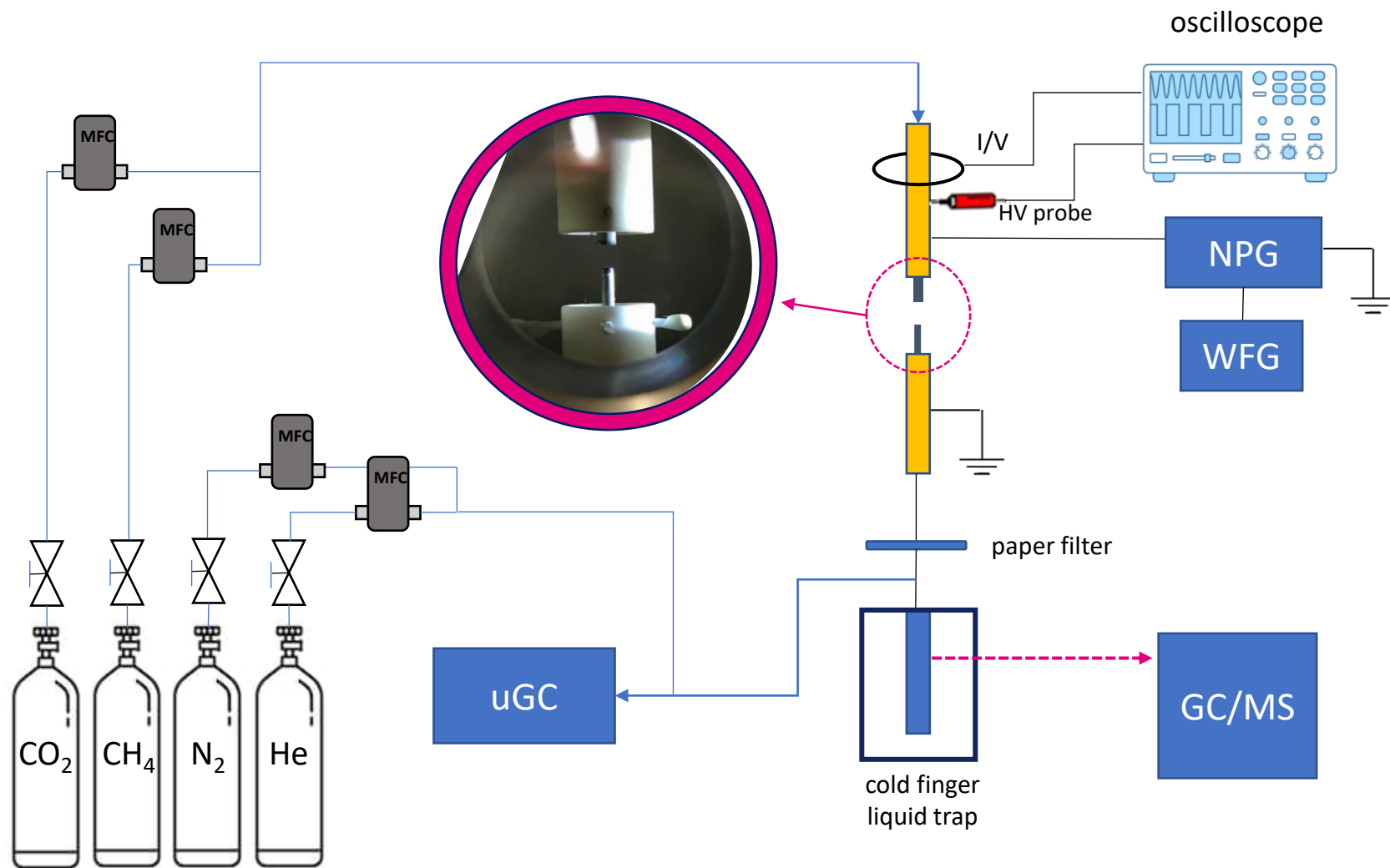
Oxidation of Ni increases with discharge energy.

◆ Ni^0 , ■ NiO , ● MgO periclase-like structure of mixed oxides

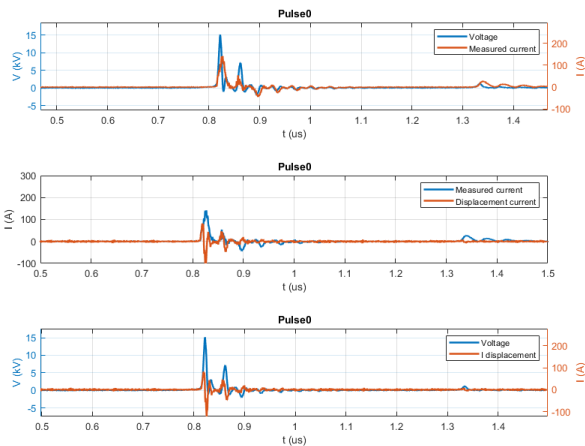
40NiHT preparation steps and spent catalyst XRD analysis



The reactor setup



An eclectic work



vacuum techniques

Study of the literature!

Chemical synthesis of catalysts

Setup of plasma discharge

Plasma-catalytic DRM

Physico-chemical material characterizations (XRD, XPS, TEM..)

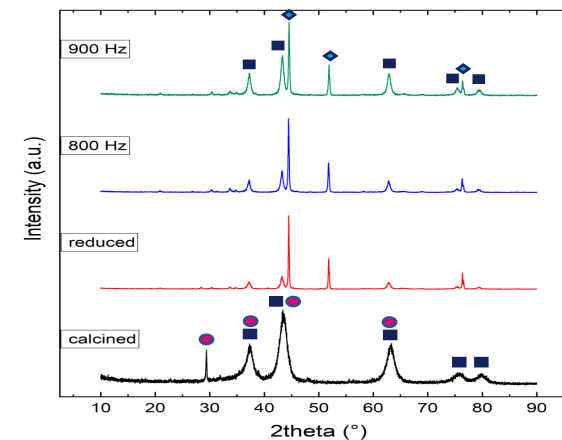
Electrical circuits

Discharge Characterization

DRM reaction with plasma and plasma + catalyst



40NiHT preparation steps and spent catalyst XRD analysis



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Thank you for your attention



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